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Windows 8, windows 7, windows Vista, windows XP, IBM Rational Rose Enterprise Edition, Rational Rose Enterprise Edition, Rational rose enterprise edition windows xp, windows 7, windows vista, windows 8 and Windows-bit, Windows-32, Windows-64 and Windows-A, Windows-8, Windows-8.1.1. Field of the Invention The present invention relates generally to the formation and use of ceramic materials and structures for use in fuel cell electrodes. More specifically, the present invention relates to fuel cell electrodes which are both highly conductive and have low electrical resistivity, and methods for their formation. 2. Description of the Related Art Fuel cells are devices which directly convert chemical energy stored in hydrocarbon fuel into electrical energy by means of an electrochemical reaction. Generally, a fuel cell comprises an anode and a cathode with an electrolyte therebetween, and fuel is supplied to the anode while air or oxygen is supplied to the cathode. The supplied fuel is oxidized at the anode and the resulting product is passed through the electrolyte to the cathode, where it is reduced. The anode and cathode are electrically coupled to the respective terminals of the cell to provide an electrical circuit between the fuel cell terminals. Fuel cell technology is of particular interest as a replacement for battery technology. The fuel cell has many advantages over the battery, including pollution-free operation, extended life and potentially higher energy density. In particular, the fuel cell can be formed into any desired shape, size and configuration, including planar sheets or disks. A plurality of fuel cells can be electrically connected to form an array of fuel cells. Because of the great interest in the fuel cell, fuel cell research has been conducted for many years. One aspect of fuel cell research has been the development of improved electrode materials. As a fuel cell is basically an electrochemical device, its performance is a function of the catalytic properties of the materials used in the fuel cell electrodes. Thus, fuel cells have been proposed and developed using a wide variety of catalytic materials. For example, fuel cells have been proposed in which the fuel is ethanol. Ethanol is a volatile, flammable liquid which is also corrosive and can interfere with the operation of the fuel cell. To provide a safer operation of fuel cells using ethanol as fuel, it has been proposed to treat the anode to increase its conductivity. For example, U.S. Pat. No. 4,654,232 to

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